

## ED-305

M.Sc. 1st Semester Examination, March-April 2021

#### CHEMISTRY

Paper - I

# Group Theory and Chemistry of Metal Complexes

<i>Time</i> : Three Hours]	[Maximum	Marks	:	80
	[Minimum Pass	Marks	:	16

**Note** : Answer **all** questions. The figures in the righthand margin indicate marks.

#### Unit-I

1.	( <i>a</i> )	Explain different types of plane of symmetry with example.	6
	( <i>b</i> )	Construct multiplication table of $C_{3\nu}$ point group.	8
	( <i>c</i> )	Explain mutual exclusion principle with example.	6
OR			

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(Turn Over)

- (2)
- (a) Explain conjugacy relation and classes. 6
- (b) The character table of  $D_3$  point group is given below. By direct product method determine the product  $E \times E$  and reduce it into the sum of irreducible representations.

$D_3$	E	$2C_3$	$3C_2$
$\overline{A_1}$	1	1	1
$A_2$	1	1	-1
Е	2	-1	0

(c) Evaluate the products  $\sigma_v$ ,  $\sigma_y$ ' and  $C_2 \sigma_v$  for a  $C_{2\nu}$  point group.

#### 6

8

#### Unit-II

<i>(a)</i>	Describe ligand group orbitals and		
	symmetry matched metal atomic orbitals		
	appropriate for $\sigma$ bonding in an octahedral ML <sub>6</sub> complex.	5	
<i>(b)</i>	Explain uses of IR Spectra to determine		
	structure of metal carbonyls.	10	
( <i>c</i> )	Explain nephelauxetic effect.	5	
OR			
( <i>a</i> )	Using MOT explain why $F^{-}$ is a weak ligand.	7	
( <i>b</i> )	Describe preparation, properties and structure of $Ni(CO)_4$ .	7	
	(b) (c) (a)	<ul> <li>symmetry matched metal atomic orbitals appropriate for σ bonding in an octahedral ML<sub>6</sub> complex.</li> <li>(b) Explain uses of IR Spectra to determine structure of metal carbonyls.</li> <li>(c) Explain nephelauxetic effect. </li> <li>OR </li> <li>(a) Using MOT explain why F<sup>-</sup> is a weak ligand.</li> <li>(b) Describe preparation, properties and</li> </ul>	

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(Continued)

## (3)

(c)	Write method	of preparation	and	structure	
	of dinitrogen	complex.			

### Unit-III

3.	( <i>a</i> )	Describe spectrophotometric method for the determination of stability constant and composition of a complex.	7
	( <i>b</i> )	Explain structure of isopoly and heteropoly acids of W.	8
	(c)	Write a short note on silicides.	5
		OR	
	( <i>a</i> )	What is chelate effect? Explain the factors affecting it.	7
	( <i>b</i> )	Describe classification of silicates with example.	7
	( <i>c</i> )	Write a short note on nitrides.	6
		Unit-IV	
4.	<i>(a)</i>	Explain structure of higher boranes.	8
	<i>(b)</i>	Explain structure of tetrameric phosphazenes.	6
	( <i>c</i> )	Write a short note on trinuclear,	
		tetranuclear metal clusters.	6
		OR	
	( <i>a</i> )	Describe method of preparation and structure of carboranes.	7
	<i>(b)</i>	Explain chain catenation and heterocatenation.	7
	( <i>c</i> )	Explain structure of borazines.	6
	C 43		720

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720

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